

# Chapter 11 Chemical Reaction Answers

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### Chapter 11 Chemical Reaction Answers

Chemistry chapter 11 Chemical reactions answer key. coefficient. a whole number that appears before a formula in an equation. spectator ion. a particle not directly involved in a chemical reaction. combustion reaction. a reaction in which oxygen reacts with another substance, often producing light or heat. reactant.

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Section 11.1 Assessment. Describe the steps in writing a balanced chemical equation. Write the skeleton equation for the following reactions: Heating solid copper(II)sulfide in the presence of oxygen gas produces pure copper and sulfur dioxide gas. Iron metal and chlorine gas react to form solid iron(III)chloride.  $\text{CuS (s)} + \text{O}_2\text{(g)} \rightarrow \text{Cu (s)} + \text{SO}_2\text{(g)}$  D

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Chapter 11 Chemical reactions. chemical equation. skeleton equation. catalyst. coefficient. A short, easy way to show a chemical reaction, using symbols. a chemical equation that does not indicate the relative amount.... a substance that initiates or accelerates a chemical reaction....

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O a combination reaction? Explain your answer. Answers 1. When two or more substances combine to form a new substance. 2.  $\text{Mg(OH)}_2$  3. No, two substances are formed. In a combination reaction, only one substance is formed. 11.5 Decomposition Reactions Practice Questions Write the reactions (including names and balanced equations) as requested on the

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Chapter 11: Chemical Reactions. a chemical change in which one element replaces a second element in a compound; also called a displacement reaction. This will only happen if the free element is higher on the activity series than the combined element is.

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Figure 11.3.2 The Relationships among Moles, Masses, and Formula Units of Compounds in the

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Balanced Chemical Reaction for the Ammonium Dichromate Volcano These are all chemically equivalent ways of stating the information given in the balanced chemical equation, using the concepts of the mole, molar or formula mass, and Avogadro's number.

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acid-base reactions: A chemical reaction that has the general form  $AB + C \rightarrow AC + B$  or  $AB + CD \rightarrow AD + CB$ ., condensation reactions : A chemical reaction that has the general form  $A + B \rightarrow AB$ .  
Condensation reactions are the reverse of cleavage reactions.

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